TEST METHODS MANUAL

Laboratory Services Division Bureau of Soils and Water Management

Dopartment of Agriculture

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- Department of Agriculture		
SECTION : SOIL CHEMISTRY	Issue No.: 3	Effective date: April 15, 2024
SUBJECT : ELECTRICAL CONDUCTIVITY OF SOIL	Revision No.: 1	Page 1 of 4

SCOPE

These methods determine the electrical conductivity (EC) of water extract at soil/water ratio of 1:1 and 1:5.

Aqueous extracts of the soil samples are usually made at higher-than-normal water content for routine characterization purposes since obtaining water samples at typical field water contents are not very practical. Considering that the amounts of various solutes are influenced by the soil/water ratio at which the extract is made, the soil: water ratio should be standardized to obtain results that can be applied and reasonably interpreted. The extraction ratios of 1:1, 1:5, etc. may be used. The 1:1 and 1:5 soil:water ratio is selected since salinity and compositional errors from dispersion, hydrolysis, cation exchange, and mineral dissolution increase as the water/soil ratio increases.

PRINCIPLE

The electrical conductivity yields a measure of the soil extract's capacity to convey an electric current. Electrical conductivity is generally related to the total solute concentration and can be used as a quantitative expression of dissolved salt concentration, even though it is also affected by the mobilities, valences, and relative concentrations of the individual ions present in the solution.

The determination of EC generally involves the physical measurement of resistance (R). The reciprocal of R is conductance (C). When the cell constant is applied, the measured conductance at a specified temperature is converted to specific C, the reciprocal of the specific R is called electrical conductivity.

Electrical conductivity increases with temperature. Conductivity ideally should be determined at 25°C. However, EC can be measured at other known temperatures and corrected to the 25°C reference using appropriate temperature coefficients (usually based on NaCl).

The presence of the major dissolved inorganic solutes, essentially Na⁺, Mg²⁺, Ca²⁺, K⁺, Cl⁻, SO₄²⁻, HCO₃⁻ and CO₃²⁻, in aqueous samples refers to the term salinity which is measured through electrical conductivity. The EC determination is often sufficient for purposes of diagnosing, surveying, and monitoring soil salinity, and for assessing the adequacy of leaching and drainage. It also, in other cases minimize the number of samples requiring compositional analyses because correlations frequently exist between salinity and the concentrations of individual solutes and their ratios within the same general area of the landscape.

TEST PRECAUTIONS

Soil samples should not be oven-dried before extracting for determination of soluble salts, because heating to 105° C converts at least a part of the gypsum (CaSO₄ · 2H₂O) to plaster of paris (CaSO₄ · 1/2H₂O). The latter hydrate has a higher solubility in water than does the former. The solubilities of other salts and minerals may also be affected.

EQUIPMENT

- a) Analytical Balance, precision of at least 0.001 g
- b) Mechanical Reciprocating shaker
- c) Electrical Conductivity meter

Prepared by:	Reviewed by:	Approved by:		
embgamboa EZRA MAE B. GAMBOA	JP Sanchy FLORFINA P. SANCHEZ	GINA PANILO, Ph.D.		
Document Controller	Head, Soil Chemistry Section	Quality Manager		
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- EC standard solution and determine again **Thank you!** Rinse the probe between samples and remo**Thank you!**